CENTEON PHARMA GMBH

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## Stabilized antithrombin III preparation

The invention relates to a stabilized antithrombin III preparation which is protected against a loss of action during pasteurization by the addition of stabilizers, and to a process for virus inactivation of an antithrombin III preparation of this type.

Antithrombin III (ATIII) is one of the most important plasmatic inhibitors. ATIII belongs to the family of serine protease inhibitors, which with their "target proteases" enter into a complex approximating the covalent bond. This complex is very stable under physiological conditions and as a rule is rapidly eliminated from the blood circulation. The reaction between ATIII and the protease is drastically accelerated by heparin, the ATIII undergoing a slight change in conformation after association with the glycosaminoglycan and thus being able to enter into an accelerated reaction with the protease. Physiologically, these processes play a role, particularly on cell surfaces which contain glycosaminoglycans, e.g. of the heparin sulfate type, and thus represent a barrier of the cells and tissue against excessive proteolytic activity. In addition, however, plasmatic coagulation, and its regulation, is also of great importance.

The regulatory function of this inhibitor is particularly clear when the ATIII plasma levels fall, as is observed in many illnesses and particularly drastically, for example, in the case of disseminated intravascular coagulation (DIC). Even a shortfall of 70% of the corresponding plasma

concentration is associated with a drastic increase in the probability of mortality. A predominance of clotting processes frequently leads to thrombotic occlusions of vessels and thus to organ failure. Correspondingly, administration of ATIII concentrates from human plasma has proven very helpful, particularly in cases of inherited and acquired deficiency states.

Patients who suffer from inherited or acquired ATIII deficiency are at present treated by substitution of concentrates obtained from human plasma. In addition to the effectiveness of these concentrates, safety with respect to the potential risk of transfer of infectious diseases must particularly be guaranteed. In addition to virus inactivation processes, such as treatment with detergents, e.g. according to the SD (solvent/ detergent) method, the heat inactivation of viruses at 60°C, which has already been used in the 1940s for albumin, is also suitable for this purpose. In general, proteins are treated at 60°C for up to 10 hours during pasteurization. This high temperature, however, can lead to denaturation of the proteins, which results in losses in activity and yield. In the case of ATIII, this is reflected in the loss of the heparin-binding properties of a specific protein fraction. This ATIII fraction is no longer measurable in a heparin cofactor test and is seen in two-dimensional immunoelectrophoresis as a separate peak, which can be clearly differentiated from the heparin-binding fraction.

The changes, which can mostly be attributed to conformational changes in the protein, are counteracted by addition of stabilizers to the solution to be heated. To achieve optimum protection of proteins of this type, the stabilizers employed for this must be exactly suited to each protein both in their composition and in the quantitative proportions of the individual constituents of the stabilizer mixture. Accordingly, US Patent 4 297 344 discloses a specific stabilizer mixture for ATIII which is used in pasteurization processes. In this patent, ATIII is mixed in aqueous solution with a carbohydrate such as sucrose and at least one amino acid from the series consisting of glycine,  $\alpha$ - and  $\beta$ -alanine, hydroxyproline, glutamine

and  $\alpha$ -,  $\beta$ - or  $\gamma$ -butyric acid. However, completely satisfactory stabilization of ATIII cannot be thus achieved, as the heparin-nonbinding fraction is over 12% after pasteurization.

It has now been found that ATIII' can be significantly more effectively protected against a loss of action during pasteurization if either one or more saccharides in a higher concentration (> 1 g/ml) or one or more saccharides in combination with one or more amino acids from the group consisting of arginine, lysine, histidine, phenylalanine, tryptophan, tyrosine, aspartic acid and its salts and also glutamic acid and its salts is selected as a stabilizer. One or more of these amino acids can also be combined with glycine and/ or glutamine here.

The stabilizers are normally employed in a buffer solution, e.g. a citrate solution.

The saccharide employed can be a monosaccharide, a disaccharide or an oligosaccharide in an amount of at least 0.5 g/ml. However, a value of over 1.0 g/ml is preferred, at the same time a pH of 6.0 to 9.5, preferably of 7.0 to 8.5, being used.

The abovementioned amino acids are employed on their own or in combination in an amount of at least 0.1 mol/l, but preferably in an amount of more than 0.5 mol/l. Combinations of one or more sugars in a concentration of more than 1.5 g/ml and one or more amino acids in concentrations of more than 0.1 mol/l in each case are particularly preferred here. The combination of a sugar in an amount of more than 1.5 g/ml and one or more of the abovementioned amino acids with glycine and/or glutamine, which are all used in concentrations of over 0.2 ml in each case, is also particularly preferred. Ammonium sulfate can be added in combination with the abovementioned stabilizers up to a final concentration of 15%.

### Example 1

5 to 10 ml each of a solution which contained ATIII in a concentration of approximately 200 IU/ml were mixed with sucrose alone and in combination with one or more amino acids and incubated at 60°C for 10 hours. The nativity of the ATIII in the sense of the heparin-binding by two-dimensional immunothen determined properties was electrophoresis. For this, an electrophoresis of the protein solution was carried out in the presence of heparin. The heparin-binding molecules became more strongly negatively charged as a result and migrated correspondingly more rapidly in the electrical field. An agarose gel with addition of a polyclonal antibody against ATIII was then poured onto the first gel and the electrical field was applied at a right angle to the direction of the first run. In regions of equimolar ratios of ATIII to antibody, a precipitation reaction took place. After soaking the gel, the precipitation line was rendered clearly visible in the form of a curve or of a peak by staining, for example, with Coomassie Blue. Quantification of the heparin-binding or -nonbinding fraction was carried out by drawing the base line with the aid of a scanner and integration of the peak areas, whose sum was set equal to 100%. The corresponding peaks were related to this and the appropriate fraction was expressed in percent. The detection limit of the method is 3% of heparin-nonbinding fraction.

Several lots were tested in each case, the ATIII solution before pasteurization always being included as a control.

The following batches were pasteurized and evaluated as described. The results are shown as mean values of 2 to 5 lots. Experiments Nos. 1, 2, 3A and 3B represent the prior art. Experiments 4, 5, 6, 7 and 8 show that in the case of an antithrombin III preparation stabilized according to the invention the heparin-nonbinding fraction is between approximately 3 and 4%.

No.	Sucrose (g/ml)	Amino acids (mol/l)	Heparin- nonbinding fraction (%)
1 2 3A 3B	before pas 0.5 1.0 1.0	teurization  glycine (2 mol/l)  glycine (2 mol/l)  (arginine (2 mol/l)/  glutamate (2 mol/l) were added only	< 3 >20 12.8
<b>4</b> 5	1.75 1.75	after pasteurization)  glycine (2 mol/l)/ glutamate (2 mol/l) glycine (2 mol/l)/ glutamate (2 mol/l)	12.6 <3.5 < 3
6 7 8	1.75 1.75 1.75	arginine (2 mol/l) lysine (2 mol/l) glutamate (1 and 2 mol/l)	4.3 3.9 < 3

#### Result:

As expected, the respective batch before pasteurization (No. 1) in each case contained <3% of heparin-nonbinding fraction ATIII. After heating, batches without, or with very low amounts of, stabilizer showed >20% of the nonbinding protein (No. 2).

The mixtures with glycine/glutamate (No. 4) with on average <3.5% and particularly with glycine/glutamate/ arginine (No. 5) with <3% showed very effective stabilizing actions. The combinations of sucrose, for example, with arginine or lysine (No. 6,7) have also turned out to be advantageous. Even glutamate on its own brought about very effective stabilization (No. 8).

To investigate whether the stabilizers added, particularly the amino acids glutamate and arginine, affected the electrophoretic running behavior on their own, these were added (No. 3B) to the sucrose/glycine mixture only after pasteurization and compared with the control (3A). It was demonstrated that these additives had no significant influence on the electrophoresis.

### Example 2:

The dependence of the stabilization on the sugar concentration was investigated according to the experimental procedure in Example 1.

No.	Sucrose (g/ml)	Amino acids (mol/l)	Heparin- nonbinding fraction (%)
1	before pas	steurization	< 3
2	0.5		>20
3A	1.0	glycine (2 mol/l) glutamate	6.8
3B	1.0	glycine (2 mol/l)	12.8
3C	1.75	glycine (2 mol/l)	5.4
4A	0.5	glutamate (1 mol/l)/ arginine (2 mol/l)	14.1
4B	1.0	glutamate (1 mol/l)/ arginine (2 mol/l)	6.4
4C	1.75	glutamate (1 mol/l) arginine (2 mol/l)	<3

#### Result:

Both the batches 3B and C and also 4A to C show that the increase in the sugar content afforded an important contribution to the stabilization. The

additionally stabilizing effect of glutamate/arginine (4B) or glutamate (3A) or compared with the batches with glycine (3A versus 3B) at a sucrose concentration of only 1 mol/l was also clear.

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### Patent claims for the USA:

- A stabilized antithrombin III preparation which is protected against a loss of action during pasteurization by the addition of stabilizers which consist of
  - a) one or more saccharides or
  - b) one or more saccharides in combination with one or more amino acids from the group consisting of arginine, lysine, histidine, phenylalanine, tryptophan, tyrosine, aspartic acid and its salts or glutamic acid and its salts, it also being additionally possible to add glycine and/or glutamine to these amino acids.
  - The stabilized antithrombin III preparation as claimed in claim 1, which, as a saccharide, contains a monosaccharide, a disaccharide or an oligosaccharide in an amount of at least 0.5 g/ml.
  - 3. The stabilized antithrombin III preparation as claimed in claim 1, which contains the saccharide in an amount of more than 1.5 g/ml.
  - 4. The stabilized antithrombin III preparation as claimed in claim 1, which contains one or more amino acids in an amount of at least 0.1 mol/l.
  - The stabilized antithrombin III preparation as claimed in claim 1, which additionally contains up to 15% of ammonium sulfate.

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- 7. as claimed in claim 2, to heat treatment at 40 to 95°C over a period of time of 5 to 50 hours.
- The process for virus inactivation of an antithrombin III preparation, 8. which comprises subjecting a stabilized antithrombin III preparation as claimed in claim 3, to heat treatment at 40 to 95°C over a period of time of 5 to 50 hours.
- The process for virus inactivation of an antithrombin III preparation, 9. which comprises subjecting a stabilized antithrombin III preparation as claimed in claim 4, to heat treatment at 40 to 95°C over a period of time of 5 to 50 hours.
- The process for virus inactivation of an antithrombin III preparation, 10. which comprises subjecting a stabilized antithrombin III preparation as claimed in claim 5, to heat treatment at 40 to 95°C over a period of time of 5 to 50 hours.

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#### Abstract:

# Stabilized antithrombin III preparation

A stabilized antithrombin III preparation is described which is protected against a loss of action during pasteurization by the addition of stabilizers which consist of one or more saccharides in a mixture with more than 0.5 mol/I of one or more amino acids from the group consisting of arginine, lysine, histidine, phenylalanine, tryptophan, tyrosine, aspartic acid and its salts or glutamic acid and its salts, it also being possible to additionally add glycine or glutamine to each of these amino acids.